

Iodine and Cancer ^[4]

A summary of the evidence to date

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Abstract

Iodine is an essential element in human physiology. Its role in thyroid function is well known and heavily weighted in the literature. Its putative role as an anticarcinogenic agent is just beginning to be widely appreciated. The molecular effects of iodine as well as ongoing epidemiological evidence points to its probable role in prevention of cancers through its antioxidant, antiinflammatory, prodifferentiating, and proapoptotic effects. This is particularly evident with stomach and breast cancers but may be relevant for many other cancers that have yet to be substantially studied.

Introduction

The first report of geographical regions with high rates of goiter having higher rates of cancer mortality was published in 1924.^{1,2} Ongoing epidemiological data has corroborated the association between goitrogenous regions and cancer incidence/mortality, particularly that of stomach cancer.^{3,4} Epidemiological evidence also suggests that thyroid disorders, particularly goiter, may be associated with breast cancer incidence and/or mortality.^{5–8} Other cancers associated with goitrogenic state include prostate cancer, endometrial, ovarian, colorectal, and thyroid cancer. It is not clear whether these associations are due to an underlying hypothyroid state, the presence of occult autoimmune processes, or iodine deficiency itself.⁹ Ultimately, the etiology of all cancers is multifactorial, with benefit assumed in the reduction of modifiable risk factors. There is substantial evidence that iodine deficiency is a modifiable risk factor in cancers of the stomach and breast and possibly many other organs. This review will outline evidence on iodine as an independent influence in cancer development and/or progression.

Total iodine content of the body is estimated at 25 mg to 50 mg, with 50%–70% of that found in extrathyroidal tissues.¹⁰ Systemic absorption of iodine takes place in the small intestine, where it is absorbed and transported into the bloodstream primarily as iodide (I⁻). Ultimately, excretion is via the kidneys with a minor amounts excreted in feces.

To enter cells, iodide (I⁻) must be cotransported with 2 molecules of sodium to overcome the electrochemical gradient. This sodium/iodide symporter (NIS) is well characterized on the basement membrane of thyroid follicular cells, where it allows for requisite iodide uptake. The means of iodide transport into enterocytes has only recently been attributed to the same symporter, NIS, that is expressed in the thyroid.¹¹ NIS found in enterocytes is controlled via a negative feedback system, such that high iodine intake reduces NIS production.¹²

Several extrathyroidal tissues also concentrate iodine via NIS found on their basement membranes. Most notably, the stomach mucosa, salivary glands, and the lactating mammary gland all have NIS that is identical to that found in the thyroid. Other tissues with high concentrations of iodine include the choroid plexus, ciliary body of the eye, lacrimal gland, thymus, skin, placenta, ovary, uterus, prostate, and pancreas.¹³ The role of concentrating iodine in lactating mammary tissue is clearly to provide necessary iodine to the developing child. The role of iodine in most other tissues is believed to include antioxidant, antiinflammatory, antiproliferative, antibacterial, proapoptotic, and prodifferentiating effects.¹⁴

Ultimately, the etiology of all cancers is multifactorial with benefit assumed in the reduction of modifiable risk factors. There is substantial evidence that iodine deficiency is a modifiable risk factor in cancers of the stomach and breast.

It is estimated that the gastrosalivary pool of iodine accounts for 23% of all iodine in the body.¹⁵ As mentioned, iodide is taken up from the bloodstream primarily via NIS found on enterocytes. Once plasma iodide reaches the mouth and stomach, it is taken up via NIS found on the basement membrane and is secreted out the apical surface to form a concentrated amount in and on the mucosa. This sets up the gastrointestinal recirculation of iodide (I⁻) that acts to conserve the overall iodine pool of the body.¹⁶ Local effects appear to contribute to the proper health and integrity of the mouth and stomach.¹⁷

While NIS allows for a high concentration of intracellular iodine in specific tissues mentioned above, iodine is believed to be in every tissue of the body.¹⁸ As early as 1961, the ubiquitous distribution of radiolabelled iodide was noted on full body imaging.¹⁹ Emerging research is showing that other receptors (such as pendrin and apical NIS) may be important in the uptake of iodine into various tissues.²⁰ In addition, iodine is liberated intracellularly when thyroxine (T₄) is converted to triiodothyronine (T₃), a process requiring the selenium dependent enzyme deiodinase.

Iodine has been proposed as a primitive antioxidant, with algae having an effective and perhaps necessary evolutionary role in squelching free radicals from the atmosphere.²¹ In humans, iodide has been shown to favorably affect serum antioxidant status.²² Iodide may be acting directly as an electron donor, squelching free radicals such as hydroxyl radicals. It may also be acting indirectly through iodination of amino acids (ie, tyrosine and histidine) or fatty acids (arachadonic acid), rendering them less likely to be oxidized themselves.²³ In an experiment to determine antioxidant capacity (ferrous reducing/antioxidant power assay, FRAP), molecular iodine (I₂) was 10 times more potent than ascorbic acid and 50 times more potent than potassium iodide (KI).²⁴ It also appears to act with thiol redox systems, such as glutathione and thioredoxin, to maintain an optimal redox balance in cells.

Iodine also has well known antiinflammatory effects.²⁵ For example, povidone-iodine has been shown to have antiinflammatory effects on wounds.²⁶ Antiinflammatory effects from iodine may be derived from actions on nitric oxide or cyclooxygenase-2.²⁷ This antiinflammatory effect has been credited with lessening the risk of peritoneal spread of cancerous cells during abdominal surgeries.^{28,29}

In concert with its antioxidant and antiinflammatory actions, iodine affects several molecular pathways that are part of differentiation and apoptosis in cells. As mentioned above, molecular species generated intracellularly through iodination of fatty acids may be significant contributors of iodine's molecular effects. One such product is 6-iodolactone (6-IL), formed from iodine (I₂) covalently bonds with arachadonic acid.³⁰ 6-IL, as well as molecular iodine (I₂), significantly influence the expression of peroxisome proliferator-activated receptor type gamma (PPAR-

gamma).³¹ PPAR-gamma is implicated in cancerous growth and development through its influences on proliferation, differentiation, apoptosis, and metabolic processes.³² One preliminary clinical study of 22 women with breast cancer given 5 mg/day of molecular iodine (I₂) found that PPAR expression was increased, along with the expected favorable effects on the cancerous tissue including increased apoptosis, decreased proliferation, and a decrease in estrogen's cellular effects.³³

In the medical literature, the generalization of the term iodine can create confusion. The term *iodine* represents essentially any form of the molecule, including molecular iodine (I₂), iodide salts (NaI or KI), iodate (NaIO₃), and/or lipids or proteins containing iodine (iodo) moieties such as iodotyrosine or iodolactones. The most well studied form is iodide salts, such as sodium iodide (NaI) and potassium iodide (KI), which are frequently used in supplementation of iodine to entire populations. Throughout this review the term *iodine* will be used as a general term, and specific forms given as a parenthetical descriptor whenever possible [eg, iodide (I⁻)].

Review of the Evidence

Breast cancer

Iodine's role in maintaining the health of breast tissue is suggested by its therapeutic effects on benign breast conditions. In a publication reviewing three clinical trials of varied designs, molecular iodine (I₂) reduced fibrocystic signs/symptoms while iodide (I⁻) was less effective and affected thyroidal function more readily.³⁴ In one of the trials included in that review, the dose of molecular iodine was 0.07 mg to 0.09 mg/kg body weight per day. Converting this to something more clinically useful, this is approximately 3.2 mg to 4.0 mg/100 lb body weight per day of molecular iodine (I₂).

In another study of 111 women with cyclic mastalgia, women took either 6 mg/day, 3 mg/day, or 1.5 mg/day of a combination iodide/iodate (I⁻/IO₃⁻), or a placebo.³⁵ Sodium iodate (NaIO₃) was used with the prediction of dissolution in the stomach to molecular iodine (I₂). In that study, more than 50% of the women taking 6 mg/day had a reduction in mastalgia symptoms at 6 months.

In keeping with iodine's effects in benign breast conditions, *in vitro* and *in vivo* studies suggest that the therapeutic form of iodine in breast cancer is molecular iodine (I₂). While NIS has been considered a necessary means for iodide uptake, human breast cancer cells (MCF-7) have been found to use facilitated diffusion of I₂ as well.³⁶ This may explain why levels of iodine are higher in cancerous breast tissue than surrounding normal tissue.³⁷ As mentioned, I₂ is capable of inducing apoptosis in human breast cancer cells through mitochondrial mediated pathways. In a rodent model of mammary carcinogenesis molecular iodine—but not iodide—was able to prevent promotion of disease.³⁸

There are several lines of evidence to support the role of molecular iodine (I₂) as preventive of carcinogenic processes. In a chemical carcinogenesis model of mammary tumors, using Sprague-Dawley rats given methyl-nitrosurea, iodine (I₂) was given as a 0.05% of water source and the rats were allowed unrestricted access. The incidence of mammary tumors was 37.5% lower in the treated rats vs controls. Further, there was an increase in proapoptotic caspase 2 and PPAR gamma expression. They also demonstrated that the vasculature of tumors in the rats given iodine (I₂) as well as vascular endothelial growth factor expression was significantly less in the tumors developed by those consuming iodine. In rats that developed tumors, there was no difference in tumor number or volume.³⁹

Iodine may also be affecting the binding of estrogen receptors to the steroid-binding element. Using breast cancer cells (MCF-7 cells), Stoddard and colleagues demonstrated that Lugol's solution (5%

iodine/10% iodide) affected 43 genes involved cell cycle growth, proliferation, and differentiation.⁴⁰ Many of the 43 genes are those upregulated by estrogens, implying that the Lugol's solution interfered with this action and had a net "antiestrogenic" effect on gene expression. This is in keeping with a rodent study using DMBA-induced mammary cancers that found a supplement of 0.1% of a combination I2/KI (0.05%/0.05%) lessened estrogen induced DNA adduct formation and increasing PPAR-gamma expression.⁴¹

Gastric cancer

The link between gastric cancers and areas of endemic goiter dates back to 1924,⁴² and ongoing data continue to corroborate this association. In a case control study of people with gastric cancer, the prevalence of goiter was more than twice that of matched controls (49.1% vs 20%).⁴³ In another case control study in Iran, urinary iodine excretion was assessed in 100 patients with gastric cancer. They found that "mean urinary iodine levels were lower in the patients with stomach cancer, 61.9 µg/g creatinine, compared to 101.7 µg/g creatinine in the control group ($P<0.0001$). More of the cancer patients (49.0%) had severe iodine deficiency (< 25 µg/g creatinine) than people in the control group (19.1%) ($P<0.0001$)."⁴⁴ In a Turkish study using direct tissue measurements, "iodine levels were lower in gastric cancer tissue (17.8 ± 3.4 ng I/mg protein) compared with surrounding normal tissue (41.7 ± 8.0 ng I/mg protein) ($P<.001$)."⁴⁵ In other studies, gastric cancers as well as later stages of Barrett's esophagus were shown to have limited or absent NIS.^{46,47} Lastly, the addition of iodine-containing salt in Poland in the 1990s to 2000s has been postulated to be responsible for the decreased incidence of stomach cancer there.⁴⁸

Its actions as an antioxidant, antiinflammatory, and prodifferentiation agent are thought to be essential to the health of cells lining the stomach.^{49,50} Systemically, iodine is recognized as an integral nutrient for proper immune function by the Institute of Medicine and as well as the United Nations Nutritional Policy Board.^{51,52} Correlations between immune deficiency, goiter, and gastric cancer have been documented in regions of Italy.⁵³ Another hypothesis of how iodine deficiency may contribute to gastric cancer comes from its local effects. In one experiment, iodine was able to inhibit carcinogenic processes involving *Helicobacter pylori*.⁵⁴

Prostate cancer

Countries such as Japan, with high iodine intake, have much lower rates of prostate cancer compared to the United States. (22.7/10,000 vs 83.8/100,000 per year, respectively).⁵⁵ While there may be other dietary and genetic influences involved, the Japanese diet is remarkably high in iodine content, with estimated intake 25 times that of the US (5,280 µg/d vs 209 µg/d).^{56,57} While intakes in the United States are relatively low, US men with the highest intake may derive some benefit, as suggested in the First National Health and Nutrition Examination Survey (NHANES I). In NHANES I, stratification of iodine excretion into tertiles showed that there was a 29% lower risk of prostate cancer in those with the highest levels vs those with the lowest [HR: 0.71 (0.51–0.9)].⁵⁸

There is ample evidence that iodine is taken up readily by the prostate and affects overall health of the organ. In animals 0.05% molecular iodine (I2) supplementation lessened symptoms of benign prostatic hyperplasia.⁵⁹ In men with BPH, 5 mg daily of Lugol's solution improved urine flow and reduced PSA values over an 8 month period.⁶⁰ In one study, NIS was found in 52% of prostate adenocarcinomas and was associated with greater aggressiveness of tumors (stage $> \text{ or } = \text{ or } = \text{pT2a}$, or Gleason $> \text{ or } = 8$).⁶¹ In another study, normal and prostate cancer cell lines were exposed to iodide, molecular iodine, and 6 iodolactone (6-IL). Unlike the normal cells, which

depended on NIS for uptake, both cancer cell lines took up I⁻ independent of NIS. Further, one prostate cancer cell line (LNCaP) was most susceptible to molecular iodine (I₂). An extension of this same study showed that iodine inhibited the growth of xenografts of prostate cancer (DU 145) in nude mice.⁶² While NIS is found in some prostate cancers, independent uptake of molecular iodine (I₂) is likely present through a facilitated diffusion process as well.⁶³

Other cancers

Molecular iodine (I₂) and 6-iodolactone (6-IL) were systematically tested on many different human cell lines, including neuroblastoma, 4 mammary cancer lines, normal mammary cells, lung, 2 glioblastoma cell lines, melanoma, 2 pancreatic cell lines, and colon carcinoma cells.⁶⁴ After a 2-day culture with molecular iodine (I₂), all cell lines were inhibited except for colon cancer. Neuroblastoma cells underwent the most complete inhibition, with MCF-7 breast cancer cells the second most sensitive. They found similar inhibitory effects were found with 6-IL. It appeared the mechanism of action included the inhibition of endothelial growth factor in this experiment. However prior experimentation by the same group found that the inhibition is most likely due to changes in the mitochondrial membrane potential, resulting in apoptosis. An effect that was completely thwarted with the addition of N-acetyl-cysteine.⁶⁵

How and whether 6-IL affects colon cancer cells is still not known. In a separate experiment, 6-IL was able to inhibit growth and induce apoptosis in colon cancer cell line (HT-29).⁶⁶ Going forward, research should clarify which cancers are most affected by I₂ and/or 6-IL.

Discussion

Historically, iodine's essential role in thyroid hormone formation and function has overshadowed its independent physiological effects. However, it is clear that iodine acts intracellularly as an antioxidant, prodifferentiating, antiinflammatory, and proapoptotic agent that is necessary for proper health and cellular renewal.

Intracellular iodinases, more specifically called iodothyronine selenodeiodinases, are essential contributors to the pool of free iodine within cells.⁶⁷ All deiodinases liberate 1 iodine (I⁻) from T₄. Type 2 deiodinase (DOI2) is found throughout various tissues and converts T₄ into T₃, resulting in the requisite loss of iodine (I⁻) into the cytoplasm. Type 3 deiodinase (DOI3), also found throughout various tissues, deiodinates T₄ but results in inactive metabolites reverse T₃ (rT₄) or 3,3'-diiodothyronine (T₂). Type 1 deiodinase (DOI1) is only found in thyroid, liver, and kidney organs and can either result in active or inactive metabolites.⁶⁸ Within each cell, the deiodinases act in concert to maintain a specific level of active T₃ that is independent of circulating levels of thyroid hormone. For example, in a hypothyroid state, DOI2 will be upregulated to increase the intracellular production of T₃ as compensation.

Normally, the various deiodinases are balanced to provide the intracellular environment with the proper amounts of T₃ and/or iodine.⁶⁹ However, they can also be manipulated within cancerous cells by the oncogenic processes. For example DOI3 is essential for the proliferation and survival of basal cell carcinoma.⁷⁰ While speculative, it is possible that reverse T₃ production serves as a means of allowing a cell to increase its intracellular pool of iodine without increasing its pool of active T₃.

The established consequences of iodine deficiency in population-based studies and the implications of iodine deficiency in the etiology of cancer certainly justify repletion of iodine in all populations through diet and/or supplementation. According to the National Institutes of Health (NIH), adults should consume a minimum of 150 mcg/day of iodine, pregnant women 220 mcg/day, and

breastfeeding women 290 mcg/day.⁷¹ The safe upper limit of consumption according to the NIH is 1,100 mcg/day for adults, with lesser amounts for teens and children. However, the NIH concedes that higher doses may be necessary for some, explaining "These [upper safe limit] levels do not apply to people who are taking iodine for medical reasons under the care of a doctor."

When assessing the physiological effects of ingested iodo-compounds, the form that is consumed is highly relevant. Most pertinent to the nutritional supplementation of iodine compounds is the finding that iodide salts (ie, KI, NaI) affect the thyroid while molecular iodine (I₂) has less influence. This is the Wolff-Cheekoff effect, which is essentially the shutdown of thyroid hormone synthesis in the presence of large amounts of iodide (I⁻). This effect is due to impaired transport of the iodide (I⁻) molecule, which is the only means for the thyroid to obtain the necessary iodine for thyroid hormone synthesis. While this effect is generally thought to be transient until the thyroid gland reequilibrates to the available iodide, there are also published studies showing a permanent impairment of thyroid function.⁷²⁻⁷⁵ Those most likely to have adverse effects were more likely to be elderly, have severe iodine deficiency, and/or have a larger increase in iodine consumption.

While the NIH recommends 1,100 mcg/day as the safe upper limit, doses up to 4.0 mg/100 lb body weight of I₂ appears to be both safe and therapeutic for benign breast diseases. The most effective anticarcinogenic form of iodine appears to be molecular iodine and intracellular 6-iodolactone. Doses exceeding 4.0 mg/100 lb of body weight have not been clinically documented as safe. In fact, there is documentation that doses 9 mg and higher may induce transient hypothyroidism as well as minor side effects such as respiratory tract infection, headache, sinusitis, nausea, acne, diarrhea, rash, or abdominal pain.⁷⁶ These side effects abated with the discontinuance of iodine, but it is important to realize that high-dose iodine is not without risk of side effects.

Many of the iodine supplements on the market provide iodide (I⁻), usually potassium iodide, either alone or in combination with molecular iodine (I₂). While this is safe for physiological repletion of iodine (doses less than 1,100 mcg/day), the salts carry greater risk of interfering with thyroid function at higher doses. The ideal supplement would contain molecular iodine with very little iodide. The difficulty is that iodine (I₂) is rendered much more soluble with iodide (I⁻) and water (ie, Lugol's solution). Given that whole food sources are generally safe, perhaps the best means of iodine repletion for prevention of thyroid disorders as well as reduction in cancer risk is through moderate consumption of whole foods such as seaweed and fish.⁷⁷⁻⁷⁹ This brings with it one difficulty—dosing. Seafoods vary dramatically in their iodo-compound type and concentration. Seaweed alone varies from 16 mcg to 2,984 mcg iodine/g depending on the type of seaweed and where it is sourced.⁸⁰

One last cautionary note, even with molecular iodine (I₂), is that patients with antithyroid antibodies can have an exacerbation of symptoms with use.^{81,82} All patients should undergo testing for autoantibodies before beginning any supplementation of iodine. In the benign breast disease studies referred to above, there were no such adverse reactions because women with autoimmune disease were excluded from the study designs.

Conclusion

Human data, including epidemiological and histological evidence along with animal and *in vitro* models, all corroborate the hypothesis that iodine compounds are essential in the health and proper differentiation of tissues. Iodine deficiency may be a hidden risk factor for cancer development and/or progression. The evidence is strongest for stomach cancer, but emerging data indicate that it may also be a risk factor in breast, prostate, and perhaps many other cancer types. There is little risk to adding whole sea vegetables to the diet. For those patients wanting to use high dose iodine

therapeutically, it is best done under the guidance of a practitioner who can monitor for any untoward effects.

About the Author



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